

M × V_L fi

Analytical vs. Actual Concentrations

Examples



Mole Relationships & Equivalence

C If the analyte is monoprotic:

$$M_t$$

$$M_{\text{NaOH}} V_{\text{NaOH}} \text{ fi } M_{\text{HCl}} V_{\text{HCl}}$$

$$M_{\text{NaOH}} V_{\text{NaOH}} \text{ fi } 2 \times M_{\text{H}_2\text{SO}_4} V_{\text{H}_2\text{SO}_4}$$