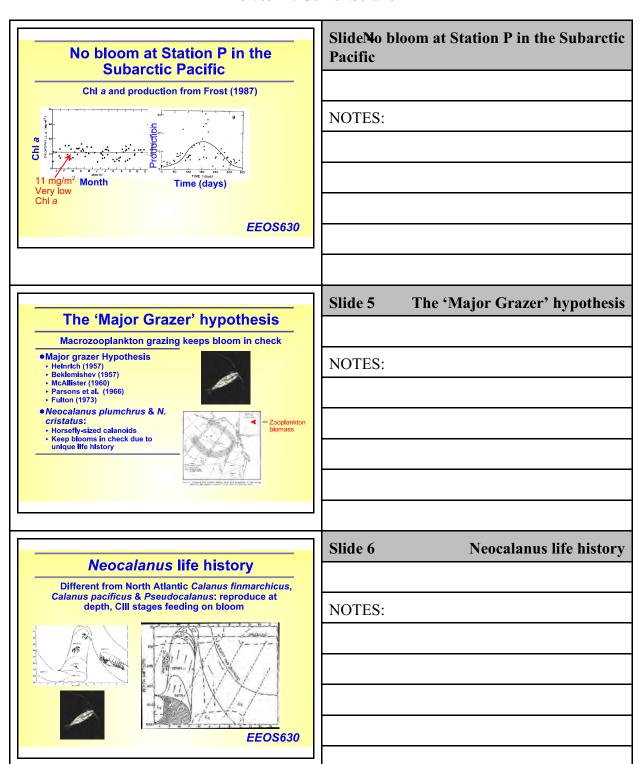
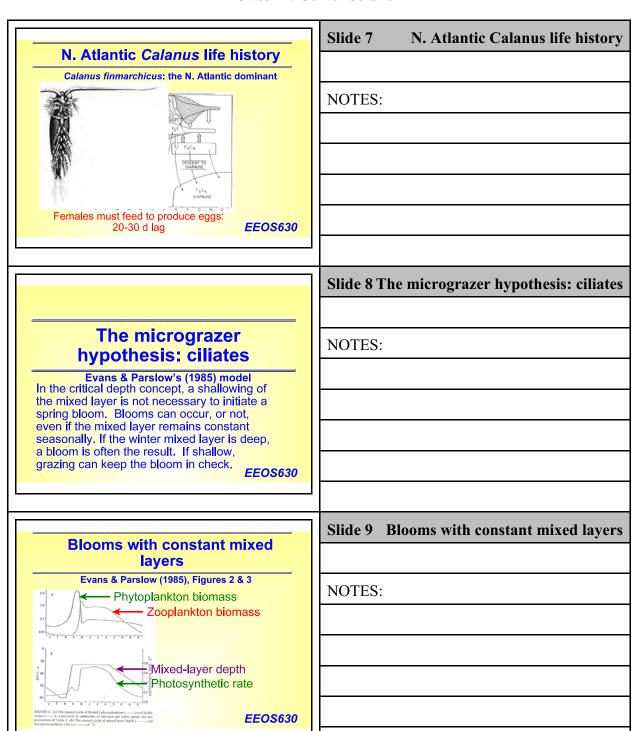
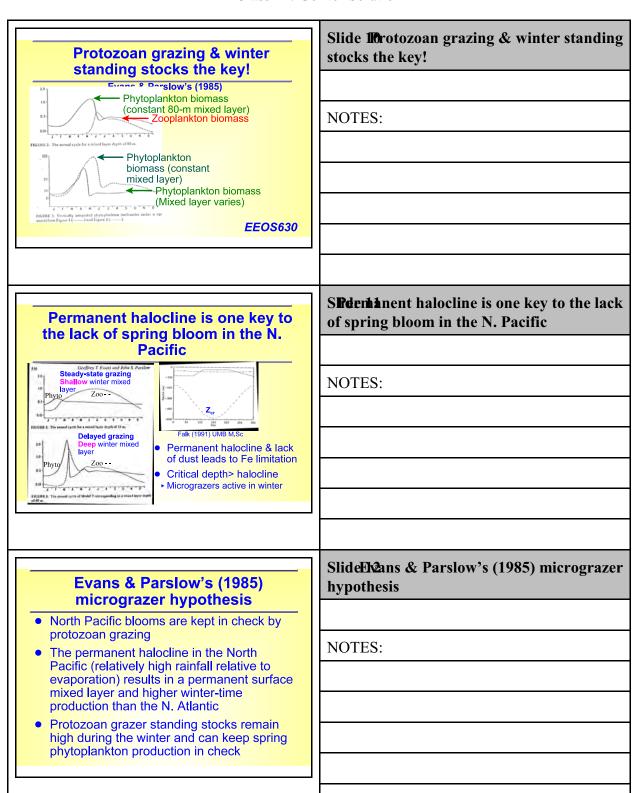
	Slide 1 The Geritol Soluti
	NOTES:
The Control Collection	
The Geritol Solution	
Class 22, Tu 18 November 2008	
EEOS630	
EEUS030	
Final evem 9 uncerning elected	Slide 2 Final exam & upcoming clas
Final exam & upcoming classes	
 Final exam will be 3-h closed book, but with essay questions provided in advance 	NOTES:
There is no set date for the final, so I'll have Angeliki send you emails to fix a time for the 3-h in class final. Copies of old final exams will be on Vista/Blackboard There is no set date for the final control of t	THOTES.
Arnab at Lowell and Christina at Amherst: when date is set, arrange for your advisor or other faculty member to proctor the	
Outer continental shelf oil effects will be due on with 1	
week left in semester. I have a set of the latest analyses being sent to me on CD by Dr. James Blake. This is a very 'hot' and current topic.	
This Thursday's class will be an introduction to zooplankton & grazing by Dr. Juanita Urban-Rich	
EEOS630	
Why no phytoplankton bloom in the Subarctic Pacific?	Slide 3 Why no phytoplankton bloom in Subarctic Pacific?
Parsons et al. (1966): the Major Grazer	
hypothesis	NOTES:
 Evans & Parslow's (1985) Micrograzer Hypothesis 	NOTES.
Martin's Iron Hypothesis	
Ecumenical Iron hypothesis	
Also consider	







Problems with the "naive" micrograzer hypothesis

- Why are there spring and fall blooms in areas like MA Bay, where the critical depth usually always exceeds the bottom depth?
- What controls diatom production, a group that owes much of its evolutionary success to its resistance to microzooplankton grazing?

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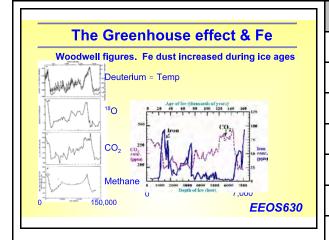
Slid P16blems with the "naive" micrograzer hypothesis

NOTES:

Martin's Geritol solution The late John Martin's hypothesis created a frenzy of activity in 1989: based on a talk at WHOI EEOS630

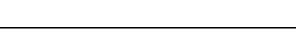
Slide 14 Martin's Geritol solution

NOTES:



The Greenhouse effect & Fe Slide 15

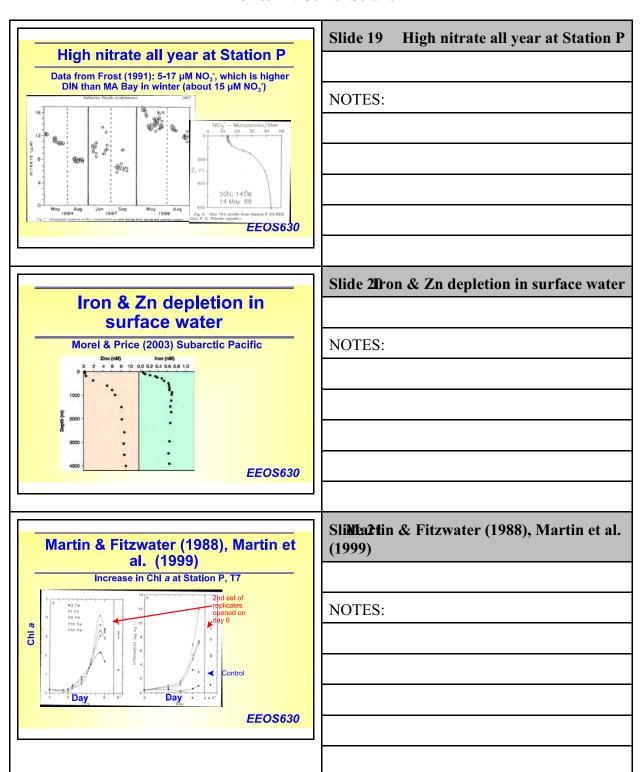
NOTES:

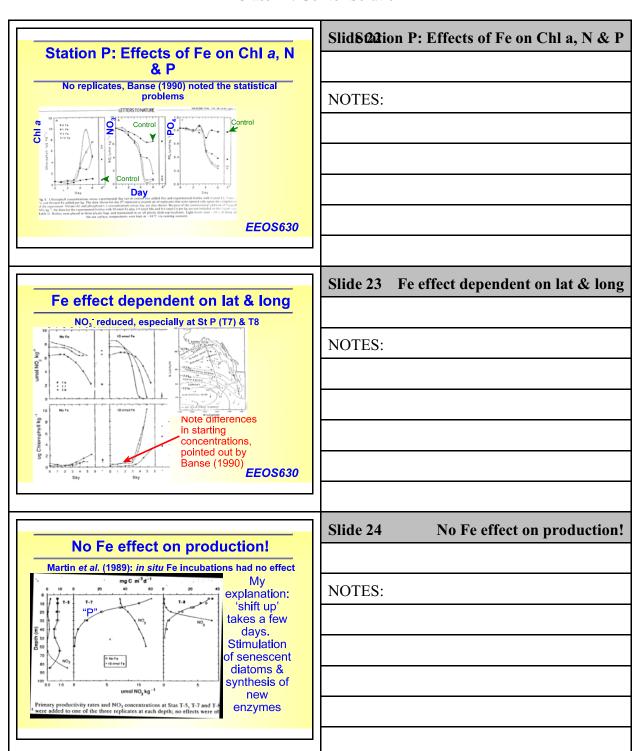


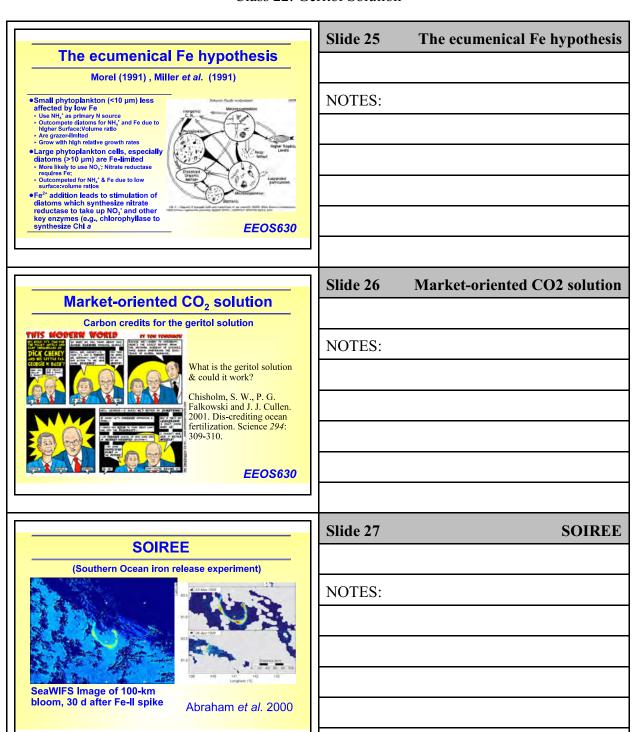
Slide 16 **Roles of Fe in plant metabolism** Roles of Fe in plant metabolism Geider & LaRoche (1994) Cytochrome oxidase NOTES: • Fe-superoxidize dismutase Catalase Peroxidase Ferrodoxin (needed for N₂ fixation) • Nitrate reductase, nitrite reductase Glutamate synthetase Others EEOS630 Slide 17 Key uses of Fe & Zn by microbes Key uses of Fe & Zn by microbes Morel & Price (2003) NOTES: Carbonic anhyd<mark>rase</mark> Nıtrate **Alkaline** reductase phosphatase Slide 18Martin & Fitzwater's Fe hypothesis Martin & Fitzwater's Fe hypothesis They argue that iron, not grazing, limit standing stocks Martin & Fitzwater's (1988) & Martin et al. (1989) sampling A NOTES: NO₃

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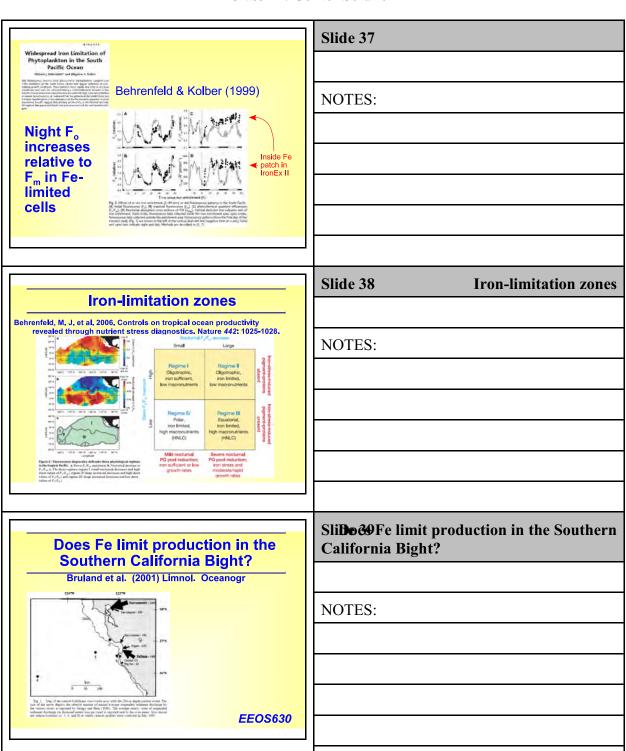




IRONEX III, SOIREE Boyd et al. (2000) Figure 2 Second 1 Day 5 Day 6 Day 10 Day 12 Day 12 Day 12 Day 12 Day 13 Day 14 Day 15 Day 16 Day 16 Day 16 Day 16 Day 16 Day 17 Day 18 Da	NOTES:
Variable fluorescence & Fe limitation Photosynthetic competency = F _v = (F _m - F _o)/F _m , Acceptor excitated Fluorescence: Photosynthetic competency =F _v = (F _m - F _o)/F _m EEOS630	Slide 29ariable fluorescence & Fe limitation NOTES:
IRONEX III: bloom by 30-50 µm diatoms Boyd et al. (2000) Fig. 3 Primary Production Primary Production	SIROSOEX III: bloom by 30-50 μm diatoms NOTES:

	Slide 31	Fe increases CO2 gradien
Fe increases CO ₂ gradient		
Watson et al. (2000): SOIREE		
Warming,	NOTES:	
[0.3° C		
But, little export!		
A STATE OF THE STA		
]	
	Slide 32	SOIREE: major result
OIREE: major results		•
ms the 'ecumenical' iron hypothesis		
in photosynthetic parameters by day 4, d by variable fluorescence	NOTES:	
chain-forming diatoms by day 5:		
l size ikton abundance quadrupled		
on small phytoplankton cells (< 20 µm cells)		
e of macrozooplankton response d carbon export to sediment traps		
ressure of CO ₂ decreased in surface		
dient would increase the ocean flux of CO ₂		
]	
	Slide 33	SOIREI
SOIREE		
al. 2000. Importance of stirring		
om 6 weeks after n experiment	NOTES:	
eaWiFS indicates the		
0.2 ±0.06 mg Chl a m ⁻³		
s a key role es of µ=0.19 d ⁻¹ orizontal diffusion=0.07 d ⁻¹		
g = 0.01 d ⁻¹ g = 0.02 d ⁻¹		
600-3000 t of		
01.0 Mg MM Mg Mg Mg		
Leading (I)	 	

Slide 34 C:N:P:Fe Redfield ratios C:N:P:Fe Redfield ratios C:N:P:Fe≈106:16:1:(0.003 to 0.0003) Lab cultures NOTES: ► Geider & LaRoche (1994) ■ Dinoflagellate (*Gymnodinium*) N:Fe ≈2000 ■ Diatom N:Fe ≈ 10,000 ■ Synechococcus (blue green) N:Fe ≈3000 ➤ Sunda et al. (1995), quoted in Fung et al. (2000) ■ Measured range N:Fe 13,000 - 116,000 ◦ Low productivity N:Fe ≈60,000 C:Fe 400,000 ◦ High productivity N:Fe ≈34,000 C:Fe 220,000 Boyd et al. (2004) Gulf of Alaska bloom ■ N:Fe 5800 C:Fe 38,000 EEOS630 Slide 35 Iron stress in the oceans Iron stress in the oceans Zones where Fe:N uptake > Fe:N supply (from dust & upwelling), Fung et al. (2000) NOTES: 1% dust-iron solubility 0% dust-iron solubility EEOS630 Slide 36 The ecumenical Fe hypothesis Controls on tropical Pacific Ocean productivity revealed through nutrient stress diagnostics NOTES: Photosynthetic competency $=F_v = (F_m - F_o)/F_m$ Nature 442: 1025-1028. August 2006



Slibod0Fe limit production in the Southern Does Fe limit production in the California Bight? Perhaps **Southern California Bight? Perhaps** Bruland et al. (2001) Limnol. Oceanogr NOTES: Diatoms (>8 μm) dominate Chl a in Fereplete regions Total Chlorophyll a (pg L 1) rangles or the Missinery Bay regard and the Hig San age of Chi o as also - 4 can live Screen Brass comm Slide 41 Problems with the Geritol solution **Problems with the Geritol solution** Not a solution for reducing atmospheric CO₂ • Fe may be the Liebigian nutrient now, but would be **NOTES:** replaced by another, e.g., Zn or Si (Leblanc et al. 2005) Increased production may not reduce the partial pressure of CO₂ sufficiently: no change in CO₂ in IronExI or IronEx II (only in SOIREE) ▶ No transport of carbon to deep waters in SOIREE Sarmiento: bottom waters, especially in the Southern Ocean, might go anoxic David Archer's calcite buffering effect: increased organic matter degradation in deep ocean sediments may dissolve calcite, increasing CO₂ concentrations: Fe only sequesters DIC on the century time scale EEOS630 Slide 42 Will Fe additions work? Will Fe additions work? http://www.nature.com/nature/journal/v407/n6805/fig_ta b/407685a0_F1.html NOTES: Short term Fe additions must decrease the CO₂ concentrations in the ocean surface There must be an enhanced flux to Long term Organic material must be sequestered in sediments Archer: Sequestration in sediments unlikely long-term Fe additions to the Southern Ocean could result in bottom-water anoxia: Sarmiento's model



Oceanography: Stirring times in the Southern

Ocean