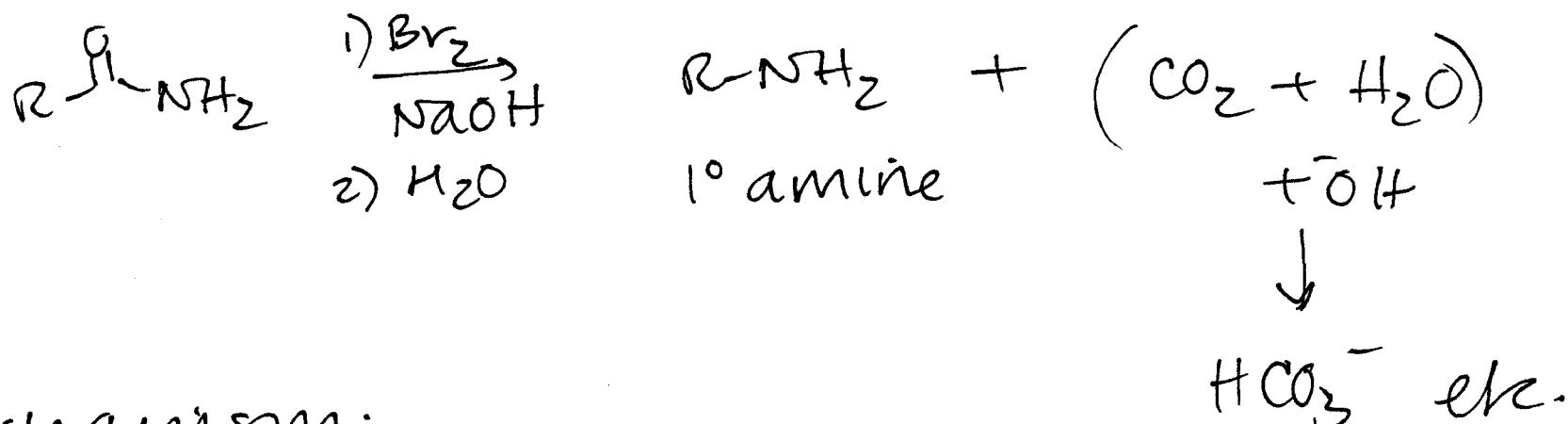
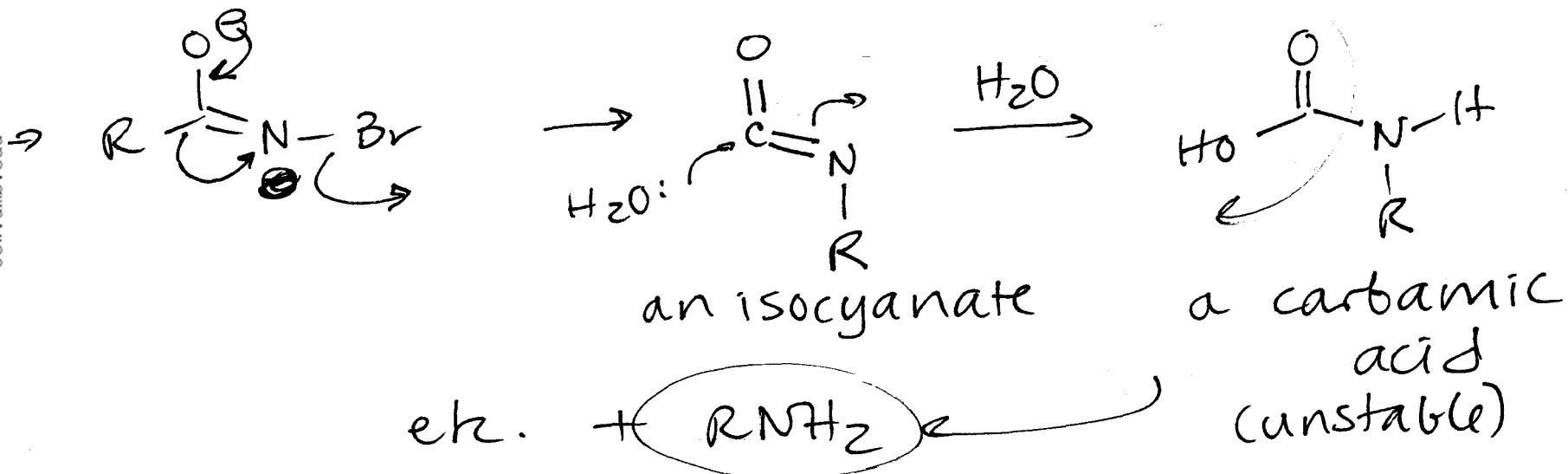
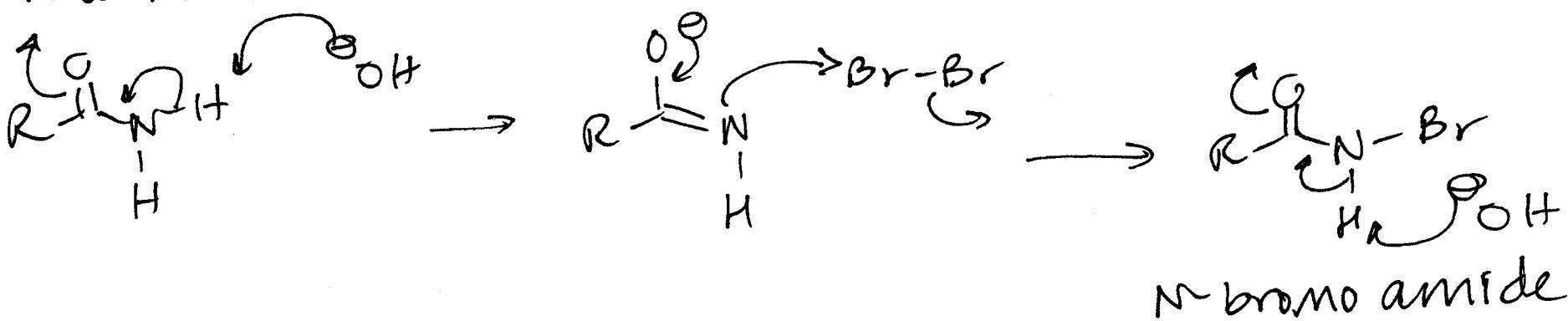


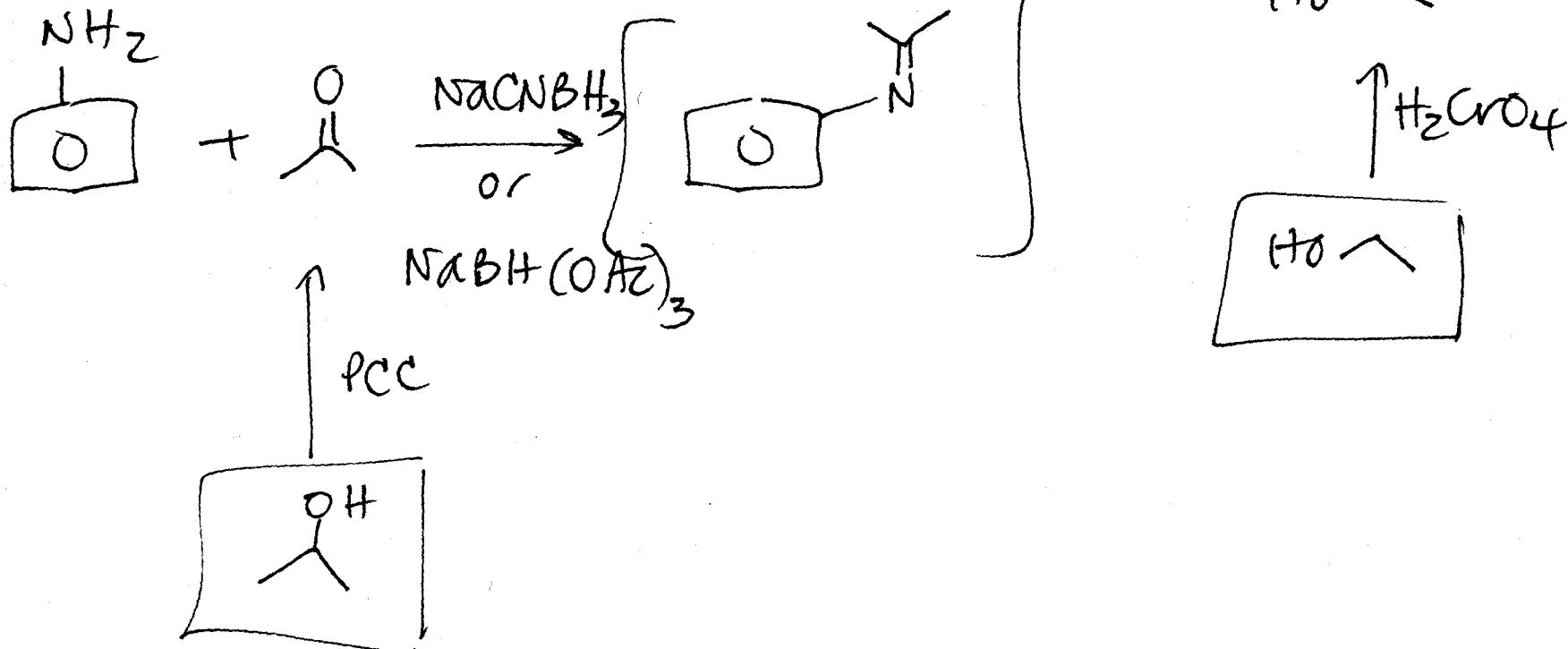
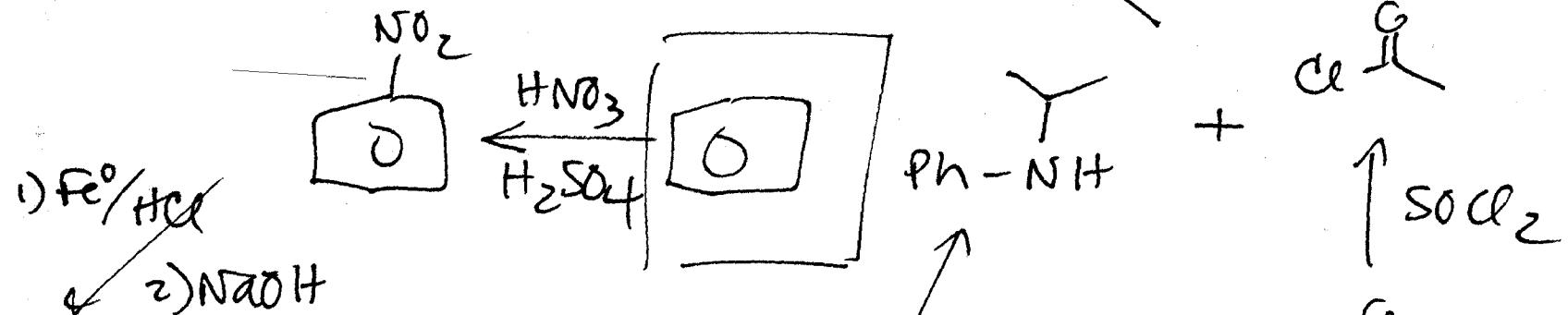
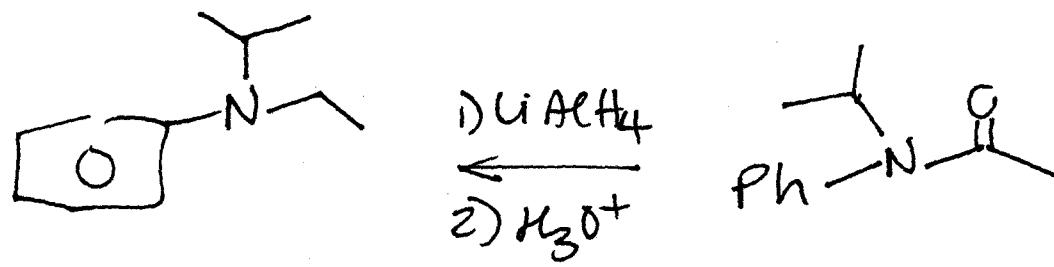
Hofmann Rearrangement - 1° amides only



mechanism:

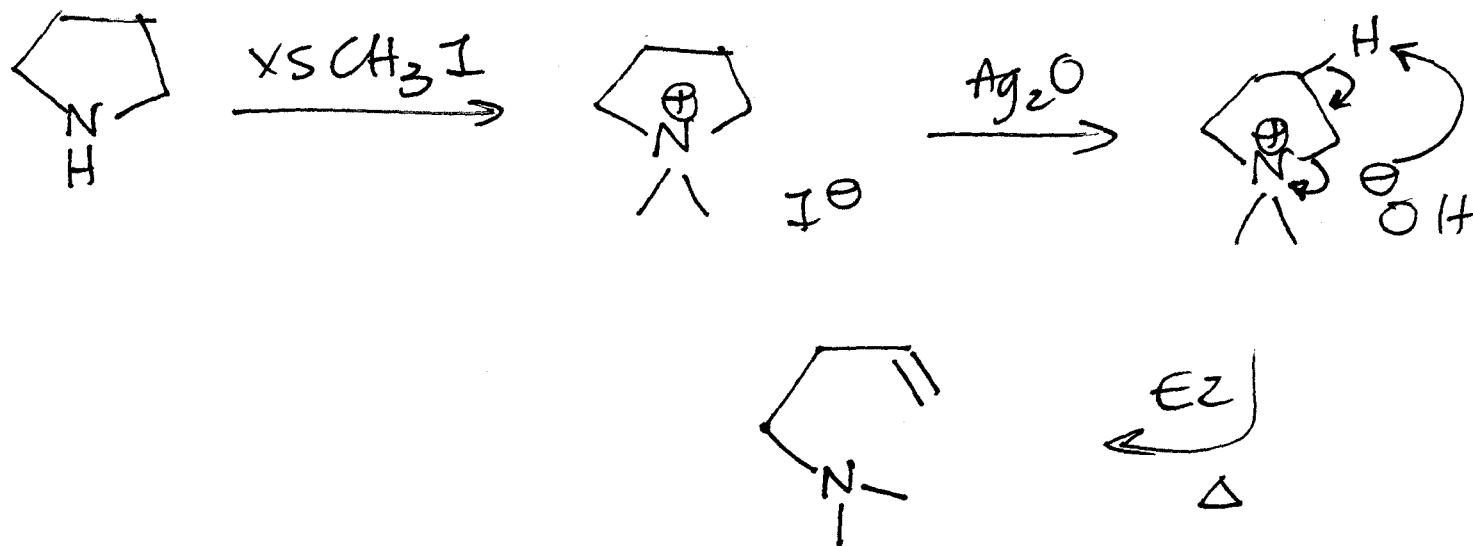


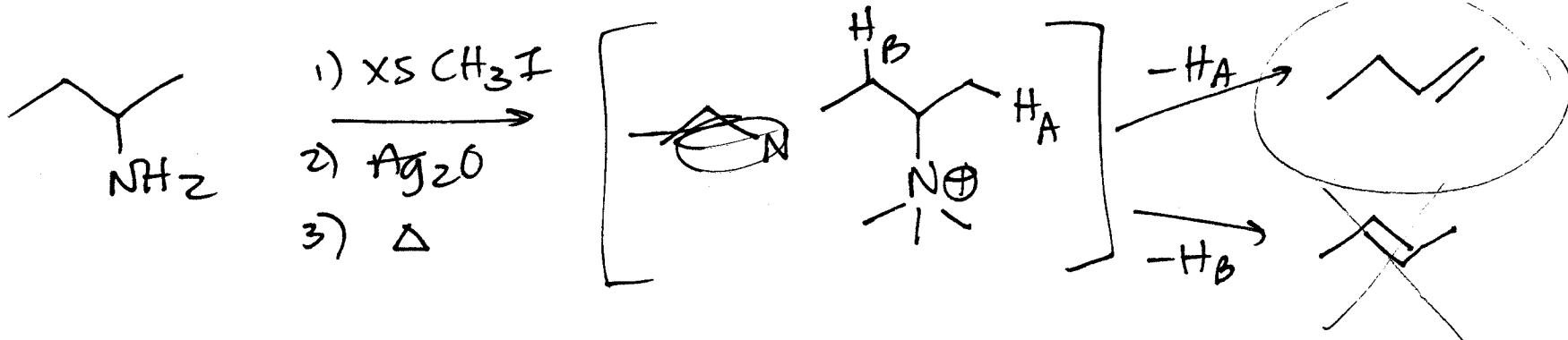
make



Reactions of Amines - bases + nucleophiles

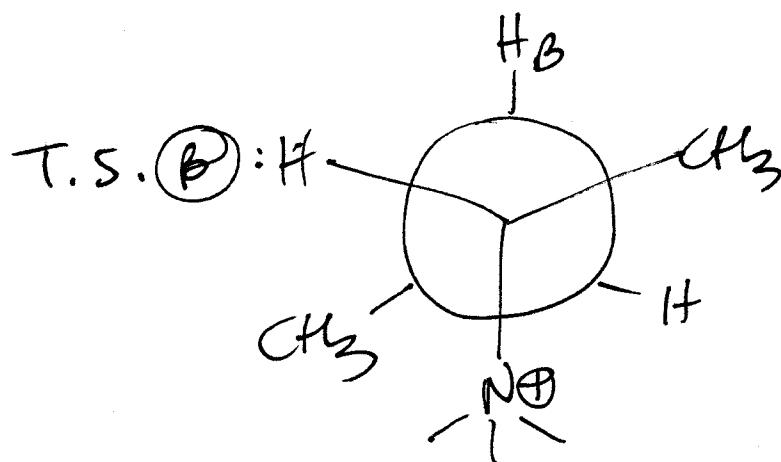
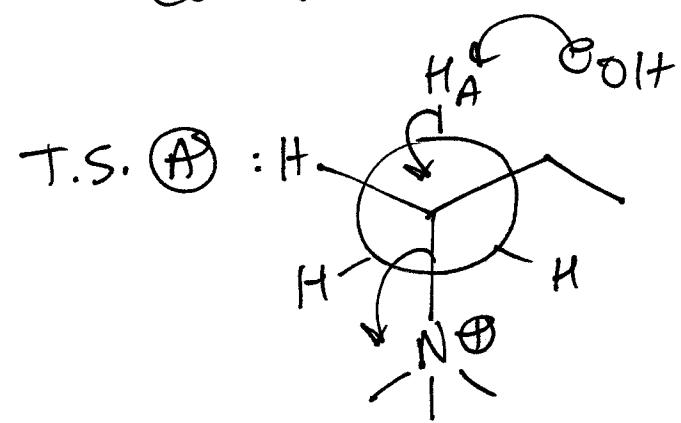
1. Rxn w/ aldehydes + ketones \rightarrow imines + enamines
2. Rxn w/ acyl chlorides, esters, etc. \rightarrow amides
3. Rxn w/ alkyl halides (S_N2) - most useful
is exhaustive methylation (react w/ $XSCl_3$)
4. Hofmann Elimination - specific to 4° ammonium salts.





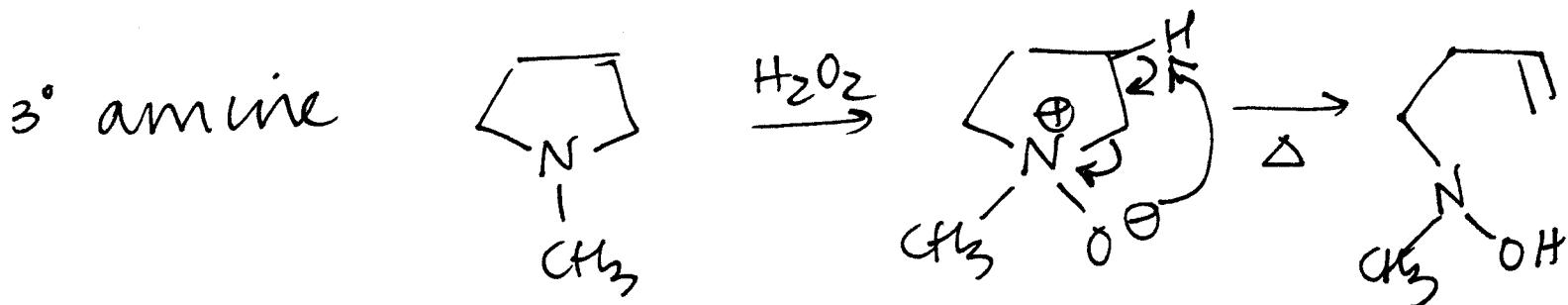
Hofmann elimination does not follow Zaitsev's rule - instead, we get the less substituted alkene.

Why? Could be electronics (see book); it could be sterics. Both give the same answer.

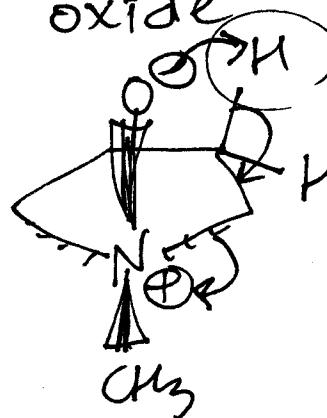


sterically less hindered

5. Oxidation of Amines - use H_2O_2 or MCPBA
 $1^\circ/2^\circ$ amines \rightarrow nothing useful

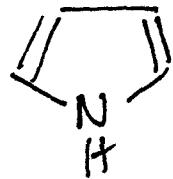


* Cope elimination
syn-elimination



6. Recall diazotization rxns

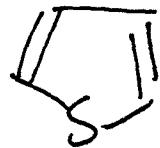
Aromatic Heterocyclic Compounds



pyrrole



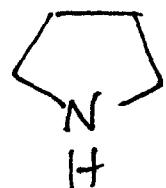
furan



thiophene



pyridine



pyrrolidine



piperidine

* these aromatics also do EAS.

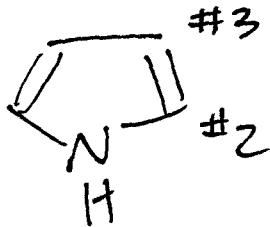
pyrrole > furan > thiophene > benzene

very
reactive

> nitrobenzene > pyridine

not very
reactive

5-membered heterocycles



EAS is preferred @ #2

(sigma complex is more stable)

6-membered heterocycles



EAS is preferred @ #3

Sigma complexes for rxn @ #2 + #4
have N w/o complete octet and w/ a δ^+

[SN_{Ar} nucleophilic aromatic substitution]

